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Biosynthesis of silver nanoparticles using *Withania somnifera* (L.) Dunal extract and its antibacterial activity against food pathogens

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Abstract

Silver nanoparticles play a significant role in the field of biology and medicine. It exerts high toxicity to various micro-organisms. The aim of the present study was to green synthesis silver nanoparticles using Withania somnifera (L.) Dunal (Ashwagandha) and evaluate its antibacterial activity against common food pathogens. W. somnifera belongs to Solanaceae (Nightshade) family. The biochemical constituent of ashwagandha plant is called withanolides. In the present study, the silver nanoparticles were synthesized in less than 30 min, using the leaf extracts of the plant, W. somnifera and its antimicrobial activity against food pathogens was evaluated. The synthesized nanoparticles exhibited antibacterial activity against E. coli and B. subtilis which was evaluated by agar-well diffusion method. The silver nanoparticles synthesized via green route were highly toxic to multidrug resistant bacteria like Pseudomonas, Staphylococcus aureus, E. coli and due to its great potential, it may be used in the biomedical application in near future.

1. Introduction

Nanoparticles have enormous applications in science and technology. Effective nanoparticles commonly used today are those made from noble metals in particular Ag, Pt, Au and Pd (Veera babu Nagati *et al.*, 2013). Compared to other nanoparticles, silver nanoparticles play a significant role in the field of biology and medicine. The nanoparticles can be synthesized by physical, chemical and biological methods but biological synthesis of silver nanoparticles have several advantages over physical and chemical methods as it is cheap, can be achieved with a single process and ecofriendly (Anal Jha and Prasad, 2010). In the biological method, living organisms such as bacteria, fungi and plants are used for the nanoparticles synthesis (Bankura and Mighty, 2012).

Silver nanoparticles are nanoparticles of silver between 1 nm and 100 nm in size. Three major sources of synthesizing silver nanoparticles were reported: bacteria, fungi, and plant extracts (Bhattacharya *et al.*, 2001). Preparation of nanoparticles by biological ways require three major elements-solvent medium for synthesis, the environmentally friendly reducer, and a nontoxic useful agent (Bharani and Thirunethiran, 2012). Several infectious diseases caused by antibiotic-resistant infective microorganism has

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brought the main focus back on the silver and its mixture forms. The properties of silver nanoparticles significant to human treatments are under investigation in laboratory and animal studies, assessing potential efficacy, toxicity, and costs (Das *et al.*, 2010).

The plant Withania somnifera (L.) Dunal commonly known as Ashwagandha or Indian ginseng belongs to Solanaceae (Nightshade) family (Elsakka et al., 1990). It contains potentially active constituents of alkaloids-isopelletierine and anaferins (Pandey et al., 2018; Su Cheol Baek et al., 2019). The biochemical constituents of Ashwagandha plant are called withanolides (Elumalai et al., 2010). These are steroidal lactones- with anolides and with aferins, saponin containing an additional acyl groups (Iuvone et al., 2003). Root of Ashwagandha is used as an anti-inflammatory drug for swellings, tumours, scrofula and rheumatism (Gupta and Kaur, 2018) and as a sedative and hypnotic in anxiety neurosis (Fidel Martinez-Gutierrez and Peggy, 2010). Ashwagandha is believed as general energy-promoting and has disease prevention property and improves the immunity (Yong Song et al., 2009). In the present study, the silver nanoparticles were synthesized in less time, using the leaf extracts of the plant, W. somnifera and its antibacterial activity against food pathogens, E. coli and B. subtilis were evaluated using agar-well diffusion method.

2. Materials and Methods

2.1 Preparation of the plant extract

W. somnifera plant material was obtained as a sample from Life Care Phytolabs Private Limited. Leaves were collected and washed 3-4 times in distilled water. Then, it is dried in shade for

7-14 days. The dried leaves are made into powder. 1 g of powder was dissolved in 100 ml of distilled water and boiled for 5-10 min at 60-70°C. The solution was filtered using Whatman's No 1 filter paper. Finally, it was collected and stored at 4°C.

2.2 Synthesis of nanoparticles

The filtered plant extract and the 10 mM of silver nitrate solution and mixed in a 250 ml conical flask in 1:4 ratio. The solution was kept in magnetic stirrer with hot plate (600°C) for 30 min for green synthesis of nanoparticle synthesis. The color change was observed visually. The supernatant mixture was centrifuged at 10,000 rpm for 15 min and the pellet was again washed in distilled water and the resultant pellet was air dried at dark condition for 2 days.

2.3 Characterization of nanoparticles

The synthesized nanoparticles were preliminary characterized using UV-visible spectroscopy. About 3 ml of the solution was taken in curettes and scanned in double-beam UV-visible spectrophotometer from 300 nm to 700 nm wavelength. The results were recorded for the graphical analysis. Further, the nanoparticles were subjected to FT-IR and Scanning electron microscopy by following Venkatesan *et al.* (2014).

2.4 Antibacterial activity

The agar well-diffusion method was used to determine the antibacterial activity of silver NPs. Different concentrations of silver NPs were tested against *E. coli*, *B. Subtilis*. Fresh bacterial suspension (10⁸ CFU/ml) was dispersed on the surface of Muller-Hinton agar plates. Different concentrations of NPs (20, 40, and 60 µl) were incorporated into the wells and the plates were incubated at 37°C for 24 h. The antibiotics were used as positive control. Zone of inhibition was measured for each plate. The diameter of the clear zone was measured using a ruler.

3. Results

3.1 Visual observation

Figure 1 shows the visual change in the colour of the Ashwagandha plant extract from greenish colour to brown colour upon addition of the silver nitrate solution, which indicates the formation of the silver nanoparticles.



Figure 1: Synthesis of silver Nanoparticles: Visual observation of the change in the reaction mixture from greenish to brown colour upon addition of silver nitrate solution.

3.2 UV-vis spectroscopy

The presence of nanoparticles was confirmed by obtaining a spectrum in the visible range of 300-700 nm. The absorption spectrum of the synthesized silver nanoparticles is shown in the Figure 2. The results revealed a peak at prominent peak at 420 nm, which confirmed the formation of silver nanoparticles.

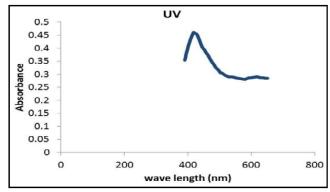


Figure 2: UV-vis spectroscopy: A prominent peak is observed at 420 nm, indicating the silver nanoparticles.

3.3 FT-IR analysis

FT-IR analysis is generally used to analyse the chemical adsorption and functional groups in the newly synthesized nanoparticles. Figure 3 shows the FT-IR transmission spectra of the Ashwagandha extract mediated silver nanoparticles. The FT-IR analysis has revealed the characteristic peaks of the silver nanoparticles.

3.4 FE-SEM analysis

Figure 4 shows the FE-SEM analysis of the silver nanoparticles, which provides a clear idea regarding the shape and size distribution of the synthesized nanoparticles. It is clearly evident from the FE-SEM micrographs that nanoparticles are evenly distributed without formation of clumps.

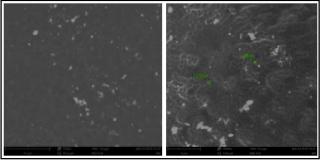


Figure 4: FE-SEM analysis of silver nanoparticles.

3.5 Antibacterial effect of silver nanoparticles

The antibacterial analysis of the synthesized silver nanoparticles are analysed by performing the agar-well diffusion experiment. Figure 5 shows the bar graph and the culture plates representing the zone of inhibition of the food pathogenic bacteria (*E. coli, B. subtilis*) upon treatment with the silver nanoparticles. The results revealed that the nanoparticles created a zone of 2.2 mm and 2.9 mm for *E.coli* and *B. subtilis*, respectively.

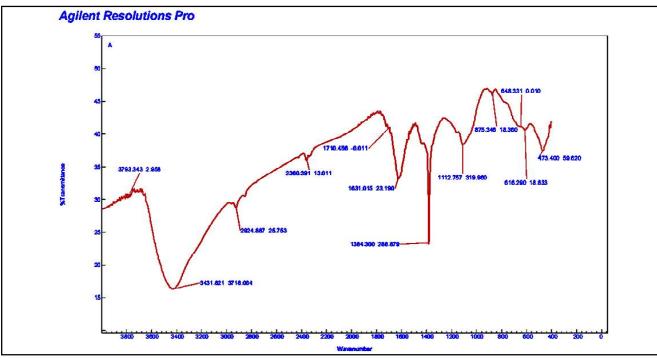


Figure 3: FT-IR analysis of the silver nanoparticles.

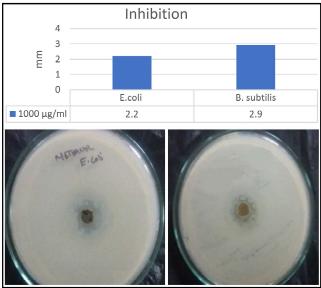


Figure 5: Antibacterial effect of silver nanoparticles: Bar graph and the representative images of culture plates of agar-well diffusion experiment for analysing the effect of silver nanoparticles on *E.coli* and *B. subtilis*.

4. Discussion

Green synthesis of nanoparticles has attained much importance due to its ease of synthesis and environmentally safe nature (Rajeshkumar and Bharath, 2017). In the present study, to synthesise plant mediated silver nanoparticles, Ashwagandha extract and silver nitrate solution were taken in 1:4 ratio and the colour changed from green to brown, which indicated the formation of silver nanoparticles. The change in

this colour is due to the reduction of the silver ion (Rajeshkumar and Bharath, 2017). This indicated the preliminary confirmation for the formation of Ashwagandha silver nanoparticles (Justin Packia Jacob *et al.*, 2012). The reduction of pure Ag+ ions was monitored by measuring the UV-Vis spectrum of the reaction medium after 30 min.

Silver nanoparticles exhibit reddish brown colour in aqueous solution due to excitation of surface plasmon vibrations in silver nanoparticles. The maximum absorbance spectrum of the silver nanoparticles was observed at 420 nm (Geethika *et al.*, 2018). Further, the results of the FE-SEM analysis revealed the nanoparticles are fairly distributed, which indicated the successful synthesis process (Kim Soo-Hwan and Hyeong-Seon Lee, 2011; Vaidyanathan *et al.*, 2010). The FE-SEM micrograph clearly showed the particle size range of our nanoparticles to be within 86 nm and high density silver nanoparticles synthesized in the presence of Ashwagandha leaf extract was further confirmed by the development of silver nanostructures (Vidhi Mehrotra *et al.*, 2011).

The nanoparticles were further subjected to FT-IR analysis which revealed the characteristic peaks of the silver nanoparticles as mentioned in the previous studies (Ziauddin *et al.*, 1996; Sushma Jain *et al.*, 2001). Further, to analyse the effect of these nanoparticles to inhibit the food pathogens, *E. coli* and *B. subtilis*; agar-well diffusion method was performed as mentioned in previous studies (Sukumaran *et al.*, 2011; Shankar *et al.*, 2003). It was clear from the results that the silver nanoparticles effectively inhibit the growth of *E. coli* and *B. subtilis*, but the effect was much prominent on *B. subtilis* compared to that of *E. coli* (Vankar and Shukla, 2012).

5. Conclusion

This study demonstrated the potential of leaf extract of the Ashwagandha plant in reducing aqueous Ag+ to Ag0 ions and formation

of ecofriendly silver nanoparticles with fairly well-defined dimensions. It provides evidence that the leaves are good source for synthesizing stable silver nanoparticles in lesser time. These ecofriendly nanoparticles could be used as an excellent source against multidrug resistant bacteria, enhancing the wound healing process that can also act as anticancer, antistress agent. It can also be used in large-scale for synthesizing nanoparticles from other inorganic materials. The silver nanoparticles synthesized *via* green route are highly toxic to multidrug resistant bacteria and due to its great potential, it may be used in the biomedical application in near future.

Conflict of interest

The authors declare that there are no conflicts of interest in the course of conducting the research. All the authors had final decision regarding the manuscript and decision to submit the findings for publication.

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